# PHENOLIC GLUCOSIDES FROM OLEA EUROPAEA SUBSP. AFRICANA

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**Key Word Index**—Olea europaea subsp. africana; Oleaceae; lignans; (+)-1-acetoxypinoresinol 4'- $\beta$ -D-glucoside; (+)-1-acetoxypinoresinol 4"-methyl ether 4'- $\beta$ -D-glucoside; (+)-1-hydroxypinoresinol 4'- $\beta$ -D-glucoside; esculin; secoiridoid; oleuropein.

**Abstract**—Two new lignan glucosides, (+)-1-acetoxypinoresinol 4"-methyl ether 4'- $\beta$ -D-glucoside and (+)-1-hydroxypinoresinol 4'- $\beta$ -D-glucoside, together with three known glucosides, (+)-1-acetoxypinoresinol 4'- $\beta$ -D-glucoside, esculin and oleuropein, were isolated from the bark of *Olea europaea* subsp. africana.

## INTRODUCTION

The bark of Olea europaea L. subsp. africana (Mill) Green (Oleaceae), known as the 'wild olive', has been used as an anti-febrile and an anti-rheumatic agent, and as a tonic in southern Africa [1]. In previous work, we reported the isolation of esculetin and scopoletin from this source [2], while Roux and co-workers isolated (+)-africanal, (-)-olivil and (+)-cyclo-olivil from the heartwood [3]. We also isolated a lignan glucoside, (+)-1-acetoxypinoresinol  $4'-\beta$ -D-glucoside, two coumarins, esculetin and esculin, and a secoiridoid glucoside, oleuropein, from the bark of the cultivated olive of commerce, O. europaea L. [4, 5].

The wild olive, previously classified as Olea africana Mill, was recently reclassified as a subspecies of O. europaea [6], and our interest was accordingly directed at the phenolic glucosides from O. europaea subsp. africana.

### RESULTS

(+)-1-Acetoxypinoresinol 4'- $\beta$ -p-glucoside (1) and (+)-1-acetoxypinoresinol 4"-methyl ether 4'- $\beta$ -D-glucoside (2) were isolated from the chloroform extract, the former being identified by direct comparison with an authentic sample [4]. The IR spectrum of 2 indicated the presence of a carboxy group (1730 cm<sup>-1</sup>) and aromatic rings (1605, 1590 and 1510 cm<sup>-1</sup>). The <sup>1</sup>H NMR spectrum of 2 exhibited resonances attributable to an aliphatic acetoxy group ( $\delta$ 1.67) and three aromatic methoxy groups ( $\delta$ 3.87). These data indicate a marked structural resemblance between 2 and 1. In the  $^{13}$ C NMR data of 2 (Table 1),  $\Delta\delta + 1.6$  (C-1"), +1.2 (C-3"), +2.0 (C-4") and -3.6 (C-5") ppm shifts of aromatic carbons relative to those of the corresponding carbons of 1, correlate with a veratryl unit in the C-6 position of the glucoside 2 [7]. The structure of 2 was confirmed as (+)-1-acetoxypinoresinol 4"-methyl ether 4'- $\beta$ -D-glucoside by its identity with the compound obtained from methylation of 1 with excess diazomethane.

Table 1. <sup>13</sup>C NMR spectral data of com-

pounds 1–3*			
Carbon	1	2	3
C-1	97.0	96.9	91.1
C-5	58.2	58.2	60.8
C-4	69.7	69.7	70.2
C-8	73.8	73.8	74.7
C-2	86.2	86.1	86.8
C-6	84.6	84.3	85.4
C-1'	130.3	130.3	131.1
C-1"	131.2	132.8	132.3
C-2'	113.0	113.0	112.6
C-2"	110.7	110.1	110.8
C-3'	148.2	148.2	148.3
C-3"	147.5	148.7	147.5
C-4'	146.3	146.3	146.0
C-4"	146.3	148.3	146.0
C-5'	114.6	114.6	114.6
C-5"	115.3	111.7	115.1
C-6'	121.1	121.0	119.7
C-6"	119.0	118.5	118.8
CH <sub>3</sub> CO	20.6	20.6	
MeCO	168.8	168.7	
MeO	55.6	55.4	55.6
	55.7	55.7	
Glc-1	99.9	99.8	100.4
Glc-2	73.2	73.2	73.2
Glc-3	76.9	76.9	76.8
Glc-4	69.7	69.7	69.7
Glc-5	76.9	76.9	76.8
Glc-6	60.7	60.6	60.8

<sup>\*</sup>The solvent used in these experiments was DMSO- $d_6$ .

(+)-1-Hydroxypinoresinol  $4'-\beta$ -D-glucoside (3), esculir (4) and oleuropein (5) were isolated from the butano extract. Compounds 4 and 5 were identified by direc

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1  $R^1 = Ac$ ;  $R^2 = H$ 

2  $R^1 = Ac$ ;  $R^2 = Me$ 

3  $R^1 = R^2 = H$ 

comparison with authentic samples [5]. Both absorption maxima in the UV spectrum of 3 and their bathochromic shifts with base were very similar to those of 1, while the <sup>1</sup>H NMR spectrum of 3 resembled that of 1 except for the disappearance of the signal assigned to an aliphatic acetoxy group. The <sup>13</sup>C NMR data indicated the absence of acetoxy resonances, but the presence of a hydroxy group at the C-1 position of 3. The structure of 3 was determined as (+)-1-hydroxypinoresinol  $4' - \beta - D$ glucoside by its identity with the compound obtained by deacetylation of 1 with ammonia in methanol. Compounds 2 and 3 are new lignan glucosides of natural origin.

Recently, we have also isolated the lignan glucosides 2 and 3 from the bark of O. europaea [8]. Thus, both subspecies africana and europaea of O. europaea are identical in their bark chemistry and have the same lignans, coumarins and secoiridoids in common [2]. It is of interest that the bark of Fraxinus japonica Blume (Oleaceae), used since ancient times as an anti-febrile and an anti-rheumatic agent in Japan [9], contains the known active principles esculetin and esculin [10]. Their common distribution also in those barks of Olea spp. which have been used in similar therapeutic applications is noteworthy.

### **EXPERIMENTAL**

All mps are uncorr. <sup>1</sup>H NMR spectra were obtained at 90 MHz using TMS as internal standard. <sup>13</sup>C NMR spectra were recorded at 15 MHz with Fourier transform using TMS as internal standard. Silica gel F<sub>254</sub> (Merck) was used for TLC. The spots were detected by spraying the plates with 10% H<sub>2</sub>SO<sub>4</sub> soln and heating.

Plant material. O. europaea subsp. africana was collected in Bloemfontein in October 1982. A voucher specimen has been deposited at the Herbarium of Higashi Nippon Gakuen University.

Isolation. Dry powdered bark (1.0 kg) of O. europaea subsp. africana was extracted  $3 \times$  with hot MeOH. The conc. extract plus  $H_2O$  was extracted successively with  $Et_2O$ , CHCl<sub>3</sub> and n-BuOH.

The CHCl<sub>3</sub> extract was chromatographed on a silica gel column using a CHCl<sub>3</sub>-EtOH gradient. The fractions were monitored by TLC developing with CHCl<sub>3</sub>-EtOH (4:1). The fractions showing a TLC spot at  $R_F$  0.30 were concd to give 1 (546.6 mg). The fractions showing a TLC spot at  $R_F$  0.43 were concd to give 2 (50.4 mg).

The n-BuOH extract was chromatographed on a silica gel column using a CHCl<sub>3</sub>-EtOH gradient. The fractions were

monitored by TLC developing with CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (65:35:10, lower layer). The fractions showing a TLC spot at  $R_F$  0.53 were concd to give 3 (34.6 mg). The fractions showing a TLC spot at  $R_F$  0.45 were concd to give 4 (132.4 mg). Those showing a TLC spot at  $R_F$  0.60 were concd to give 5 (7.7 g).

(+)-1-Acetoxypinoresinol 4'-β-D-glucoside (1). Colourless needles, mp 183.5–185°, [α] $_{\rm D}^{122}$  + 7.9° (c 1.0; EtOH). UV  $\lambda_{\rm max}^{\rm EIOH}$  nm (log ε): 231 (4.31), 279.5 (3.83). UV  $\lambda_{\rm max}^{\rm EIOH}$  + NaOH nm: 254, 280, 292. IR  $\nu_{\rm max}^{\rm KBr}$  cm $^{-1}$ : 3325 (OH), 1735 (CO), 1600, 1590, 1520 (aromatic C=C). <sup>1</sup>H NMR (CD<sub>3</sub>OD): δ1.67 (3H, s, alcoholic OAc), 2.77–3.03 (1H, m, H-5), 3.87 (6H, s, 2 × MeO), 3.24–4.44 (4H, m, H-4, 8), 5.05 (1H, s, H-2), 6.67–7.30 (6H, m, Ar-H). (Found: C, 57.68; H, 5.93. C<sub>28</sub>H<sub>34</sub>O<sub>13</sub>·1/2H<sub>2</sub>O requires: C, 57.23; H, 6.00%).

(+)-1-Acetoxypinoresinol 4"-methyl ether 4'-β-D-glucoside (2). Amorphous powder,  $[\alpha]_D^{20} + 9.1^\circ$  (c 1.6; EtOH). UV  $\lambda_{\max}^{EtOH}$  nm (log ε): 231 (4.22), 279 (3.72). IR  $\nu_{\max}^{KBr}$  cm<sup>-1</sup>: 3400 (OH), 1730 (CO), 1605, 1590, 1510 (aromatic C=C). <sup>1</sup>H NMR (CD<sub>3</sub>OD): δ1.67 (3H, s, alcoholic OAc), 2.75–3.08 (1H, m, H-5), 3.87 (9H, s, 3 × MeOH), 3.22–4.44 (4H, m, H-4, 8), 5.05 (1H, s, H-2), 6.77–7.28 (6H, m, Ar-H). (Found: C, 57.53; H, 5.97. C<sub>29</sub>H<sub>36</sub>O<sub>13</sub>·H<sub>2</sub>O requires: C, 57.04; H, 6.27%)

Compound 1 in MeOH was methylated with excess  $CH_2N_2$  to give (+)-1-acetoxypinoresinol 4"-methyl ether 4'- $\beta$ -D-glucoside, which was identical to 2 in all respects.

(+)-1-Hydroxypinoresinol 4'-β-D-glucoside (3). Colourless plates, mp 127–129°, [α] $_{\rm D}^{23}$  – 9.3° (c 0.42; MeOH). UV  $\lambda_{\rm max}^{\rm EIOH}$  nm (log s): 228 (4.11), 279.8 (3.66). UV  $\lambda_{\rm max}^{\rm EIOH}$  +NaOH nm: 254, 280, 292. IR  $\nu_{\rm max}^{\rm KBr}$  cm<sup>-1</sup>: 3400 (OH), 1600, 1515 (aromatic C=C). <sup>1</sup>H NMR (CD<sub>3</sub>OH): δ 2.96–3.14 (1H, m, H-5), 3.80, 3.83 (6H, each s, 2 × MeO), 3.47–4.52 (4H, m, H-4, 8), 4.80 (1H, s, H-2), 6.60–6.80 (6H, m, Ar–H). FDMS m/z: 536 [M] $^+$  (C<sub>26</sub>H<sub>32</sub>O<sub>12</sub>).

Compound 1 was deacetylated with NH<sub>3</sub> in MeOH to give (+)-1-hydroxypinoresinol 4'- $\beta$ -p-glucoside, which was identical to 3 in all respects.

Esculin (4). Colourless crystalline powder, mp  $152-154^{\circ}$ . UV  $\lambda_{\text{max}}^{\text{EiOH}}$  nm (log s): 224 (4.12), 250 (3.67), 298 (3.79), 336 (4.07). IR  $\nu_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup>: 3100–3550 (OH), 1700 (CO), 1680 (C=C), 1610, 1570 (aromatic C=C).  $^{1}$ H NMR (CD<sub>3</sub>OD + DMSO- $^{4}$ 6):  $\delta$ 6.27 (1H,  $^{4}$ ,  $^{4}$  = 10 Hz, H-3), 6.87, 7.47 (2H, each s, Ar-H), 7.87 (1H,  $^{4}$ ,  $^{4}$  = 10 Hz, H-4).  $^{13}$ C NMR (DMSO- $^{4}$ 6):  $\delta$ 160.7 (C-2), 112.1 (C-3), 144.5 (C-4), 114.7 (C-5), 142.7 (C-6), 151.4 (C-7), 103.2 (C-8), 150.5 (C-9), 110.9 (C-10), 102.3 (C-1'), 73.4 (C-2'), 77.3 (C-3'), 69.9 (C-4'), 76.1 (C-5'), 60.8 (C-6'). (Found: C, 51.65; H, 4.80: C<sub>15</sub>H<sub>16</sub>O<sub>9</sub> · 1/2H<sub>2</sub>O requires: C, 51.58, H, 4.91 %).

Oleuropein (5). Amorphous powder,  $[\alpha]_D^{22} - 128.4^\circ$  (c 0.61; EtOH). UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm (log ε): 228 (4.25), 280 (3.74). IR  $\nu_{\text{max}}^{\text{KBr}}$  cm<sup>-1</sup>: 3200-3550 (OH), 1700 (CO), 1620 (C=C), 1590, 1510 (aromatic C=C). <sup>13</sup>C NMR (CD<sub>3</sub>OD): δ95.0 (C-1), 154.9 (C-3), 109.1 (C-4), 31.5 (C-5), 41.0 (C-6), 172.9 (C-7), 124.6 (C-8), 130.3 (C-9), 13.2 (C-10), 168.5 (C-11), 100.7 (C-1'), 74.5 (C-2'), 78.1 (C-3'), 71.2 (C-4'), 77.7 (C-5'), 62.5 (C-6'), 130.5 (C-1"), 116.2 (C-2"), 146.0 (C-3"), 144.7 (C-4"), 116.8 (C-5"), 121.0 (C-6"), 68.6 (C-α), 35.1 (C-β), 53.0 (MeO). (Found: C, 52.21; H, 6.04. C<sub>25</sub>H<sub>32</sub>O<sub>13</sub>· 2H<sub>2</sub>O requires: C, 52.08; H, 6.29 °<sub>0</sub>.)

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